

MORPHOLOGICAL CHANGES IN CARBON-NANOTUBE/SILICA COMPOSITES AT DIFFERENT MATRIX POROSITIES

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Abstract

Carbon Nanotubes (CNTs) have become very attractive fillers in various host materials, commonly in organic polymers, due to their ability to enhance electrical [1-2], thermal and mechanical properties [1] and moreover, they have shown to possess nonlinear optical properties [2-3]. However, electro-optical applications of organic polymers are limited since they lack optical quality and thermal stability. In contrast, glassy matrices have favorable properties for electro-optical applications. Sol-gel technology is a novel method for fabrication of glassy materials at temperatures below 100°C with controllable porosity, from highly porous silica to completely sealed glass [4-5]. A major issue in CNTs composite fabrication is the ability to achieve homogeneous dispersion of the CNTs within the matrix, since CNTs tend to form aggregates. This problem poses even a greater challenge in the case of sol-gel glasses, as the CNT dispersion must occur simultaneously with gelation.

In this work we present preparation methods and some characterization of MWNT/Aerogel, MWNT/Xerogel and MWNT/Fast-Sol-Gel composites. The sol-gel technology allows fabrication of various glassy materials with variable porosity regulated by the fabrication procedure. By using a slow drying procedure at ambient conditions, semi porous silica, named Xerogel, is obtained (Figure 1a). A supercritical drying results in highly porous silica, named Aerogel (Figure 1b). While using the Fast-Sol-Gel process [6], which involves rapid drying with, elevated temperature and pressure, a non porous matrix is obtained (Figure 1c). The aforementioned materials have a characteristic surface morphology, as can be seen from Environmental Scanning Electron Microscopy (E-SEM) images given in Figure 2.

The Xerogel has a typical grained surface with small pores of about 2-3 nm diameter (Figure 2a). The Aerogel matrix, demonstrates a sponge-like surface with pores of about 20 nm diameter (Figure 2b). While the Fast-Sol-Gel matrix possesses a flat and dense surface without notable porosity (Figure 2c).

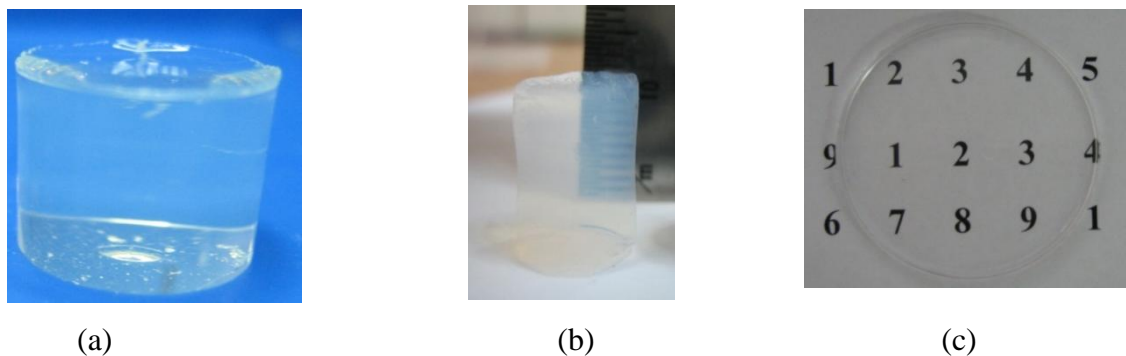


Figure 1. Optical images of glassy materials prepared by sol-gel technology differing in their porosity: (a) Xerogel, a semi porous matrix, (b) Aerogel, a highly porous matrix, (c) Fast-Sol-Gel, a hybrid organic-inorganic non porous matrix.

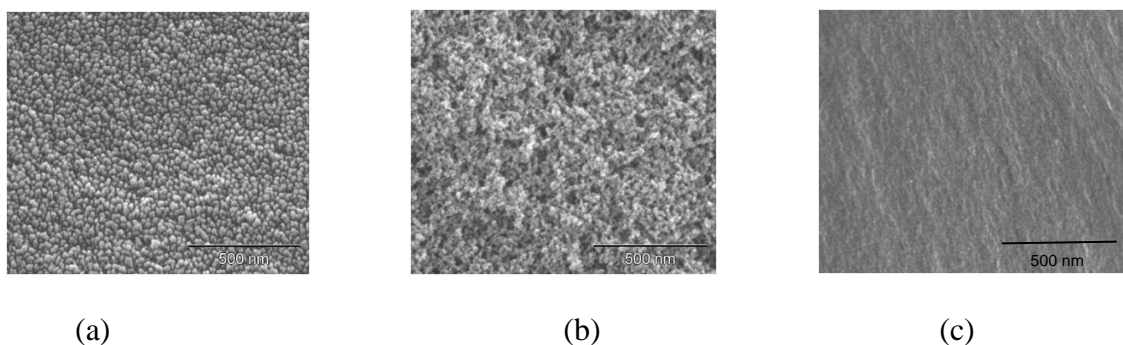


Figure 2. E-SEM images of glassy materials prepared by sol-gel technology with different porosity: (a) Xerogel, grained surface, (b) Aerogel, sponge-like surface, (c) Fast-Sol-Gel, smooth surface. The scale bar for all figures is 500 nm.

CNT/Silica composites with varying porosity were prepared by the sol-gel routes described above. We have successfully prepared CNT/Xerogel (Figure 3a), CNT/Aerogel (Figure 3b), and CNT/Fast-Sol-Gel composites (Figure 3c). For each composite we adapted a different approach to incorporate the CNTs. Finally, to characterize the morphology of the new composites we used gas adsorption analysis and electronic microscopy imaging. From the morphological results we found a correlation between the CNT concentration and the glass porosity. These aspects and more will be discussed in more details in the talk.

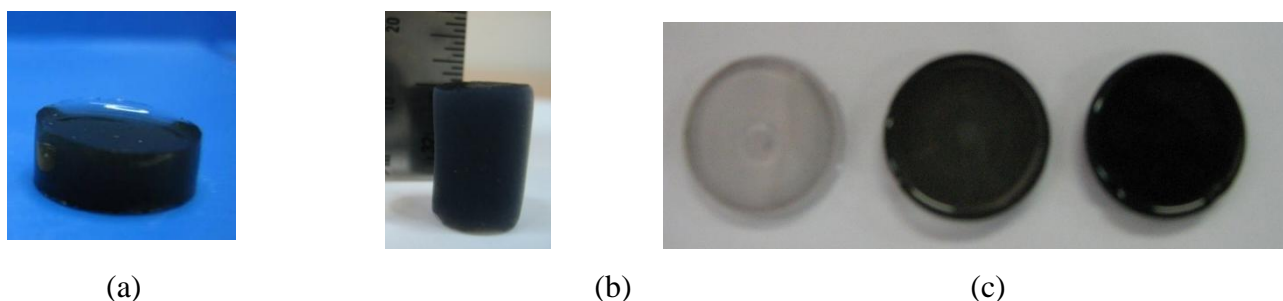


Figure 3. Optical images of CNT/Silica composites by sol-gel technology differing in their porosity: (a) CNT/Xerogel, (b) CNT/Aerogel, (c) CNT/Fast-Sol-Gel, with various CNT concentrations.

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